

Note

X-Ray structure of 3,4,5-tri-*O*-acetyl-2,6-anhydro-D-mannonitrile (2,3,4-tri-*O*-acetyl- α -D-arabinopyranosyl cyanide)

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In connection with work on a new method for the synthesis of glycopyranosyl cyanides¹, the crystal structures of four such 2,6-anhydroaldonitriles have been described². The structure of 3,4,5,7-tetra-*O*-acetyl-2,6-anhydro-D-*glycero*-L-*manno*-heptonitrile (2,3,4,6-tetra-*O*-acetyl- β -D-galactopyranosyl cyanide) was published earlier³. We now describe, as another example, the conformation which is adopted by the title cyanide **1**⁴ in the crystalline state.

Compound **1** is homomorphous to the L enantiomer of the above *galacto* compound. From ¹H-n.m.r. measurements¹, it was deduced that, in solution in chloroform, **1** populates the alternative ⁵C₂(D) and ²C₅(D) conformers in the ratio 48:52. In benzene as solvent¹, the calculated amount of the ²C₅(D) conformer increased to 67%. Thus, the solid state conformation of **1** was not predictable readily from these results and, therefore, the X-ray structure was determined.

3,4,5-Tri-*O*-acetyl-2,6-anhydro-D-mannonitrile (**1**) crystallises in the orthorhombic space-group *P*2₁2₁. Further crystallographic information is given in Table I. The structure was solved by direct methods using the computer program

TABLE I

CRYSTALLOGRAPHIC DATA FOR **1**

Formula	C ₁₂ H ₁₅ NO ₇	Density (calc.) (g, cm ⁻³)	1.30
Mol. wt.	285.25	λ (MoK α) (pm)	71.069
M.p. (degrees)	136–138 ⁴	μ (cm ⁻¹)	0.70
Space group	<i>P</i> 2 ₁ 2 ₁	2 θ range (degrees)	3–52
Cell constants (pm)		Reflections measured (one octant only)	1500
<i>a</i>	645.6(1)	Reflections with $F_0 > 4\sigma(F_0)$	1093
<i>b</i>	1121.9(1)	Number of refined parameters	217
<i>c</i>	2011.6(1)	Final residual factors	
Volume (pm ³)	1457.03(1) $\times 10^6$	<i>R</i>	0.051
<i>Z</i>	4	<i>R</i> _w	0.054
<i>F</i> (000)	600	Diffractionmeter	Siemens AED2

TABLE II

FRACTIONAL POSITIONAL PARAMETERS ($\times 10^4$) AND TEMPERATURE FACTORS U_{eq} ($\times 10^3$) OF CARBON, NITROGEN, AND OXYGEN ATOMS IN **1**

Atom	x	y	z	U_{eq}
O-2	2394(5)	1923(2)	6658(2)	54(2)
O-3	1410(6)	504(3)	5528(2)	67(2)
O-4	2590(6)	-1664(3)	6069(2)	67(2)
O-5	5603(6)	-697(3)	6948(2)	77(2)
O-21	-609(6)	1462(4)	7136(2)	97(3)
O-31	3097(11)	1242(5)	4661(2)	124(4)
O-41	3121(8)	-2927(3)	5226(2)	88(3)
N-11	4278(10)	919(5)	8245(3)	97(4)
C-1	4906(8)	493(5)	6997(3)	62(3)
C-2	2910(8)	680(4)	6608(2)	50(3)
C-3	3337(9)	391(4)	5880(2)	55(3)
C-4	4180(10)	-864(5)	5823(3)	68(3)
C-5	6056(11)	-998(6)	6259(4)	88(5)
C-11	4568(9)	718(5)	7698(3)	72(4)
C-21	555(9)	2200(5)	6941(2)	59(3)
C-22	198(10)	3512(5)	6965(3)	86(4)
C-31	1500(13)	942(4)	4911(3)	76(4)
C-32	-568(14)	989(6)	4601(3)	99(5)
C-41	2248(9)	-2673(4)	5725(3)	60(3)
C-42	656(9)	-3405(4)	6067(2)	69(3)

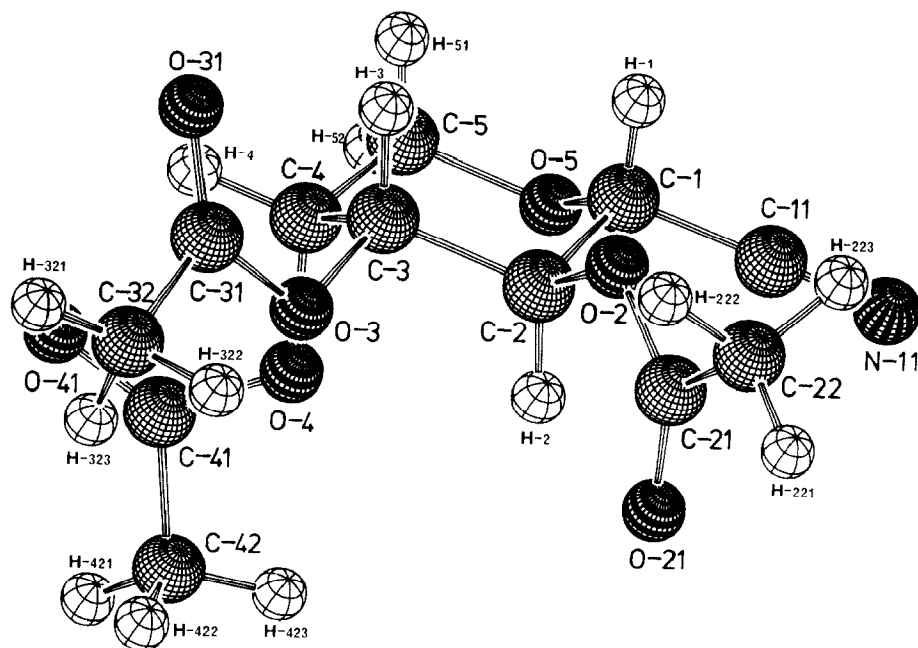


Fig. 1. A SCHAKAL⁷ drawing of a molecule of **1** showing atom numbering.

SHELXS-86⁵ for the phase problem, and the program SHELX-76⁶ (full matrix) for refinement. The final positional parameters (temperature factors U_{eq} included) of the carbon, nitrogen, and oxygen atoms are listed in Table II*.

A SCHAKAL representation⁷ of one molecule of **1** is shown in Fig. 1; the numbering follows the glycosyl cyanide nomenclature.

Compound **1** adopts the ${}^2C_5(D)$ conformation (1C_4 when viewed as a glycosyl cyanide). The analogous enantiomeric conformation is found in the related acetylated β -D-galactopyranosyl cyanide³ due to the bulky exocyclic acetoxymethyl group. The "puckering" parameters⁸ of the pyranose ring in **1** are $Q = 58.5$ pm, $\phi = 268.6^\circ$, and $\theta = 177.6^\circ$, which are indicative of an only slightly distorted chair. The respective values for the β -D-galactosyl compound³ (L enantiomer) are $Q = 56.9$ pm, $\phi = 185.5^\circ$, and $\theta = 172.5^\circ$. The orientation of the acetoxy groups in **1** is also very similar to that found in the homomorphous *galacto* cyanide³. Both compounds thus follow trends which are generally observed in carbohydrate acetates (*cf.* the discussion in ref. 1).

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*Lists of observed and calculated structure amplitudes, atomic co-ordinates of the hydrogen atoms (calculated in a late stage of refinement), anisotropic thermal factors U_{ij} for C, N, and O, and Tables of bond lengths and bond angles are deposited with, and can be obtained from, Elsevier Science Publishers B.V., BBA Data Deposition, P.O. Box 1527, Amsterdam, The Netherlands. Reference should be made to No. BBA/DD/401/*Carbohydr. Res.*, 184 (1988) 247–249.